

Course Code	Course Name	Credits
26PH604	PHYSICAL-TECHNICAL IN MATERIAL SCIENCE	04

### Course Objectives

- To impart knowledge on various materials, smart materials, optical composite, ceramic and superconductors' techniques.
- To learn the structural, ceramic and glass making, properties of materials, and surface characterization techniques.

### Learning Outcomes

Upon successful completion of this course it is intended that a student will be able to:

- Gain information about atomic structure, atomic bonds, crystal structure, crystal geometry and crystal defects

### Unit 1 – Phase Transition In Materials (13 Hrs.)

Definition and basic concepts – solubility limit – phases – microstructure – phase equilibria – unary phase diagrams – Binary phase diagrams – Binary isomorphous systems – Interpretation of phase diagrams – Development of microstructure in isomorphous alloys – mechanical properties of isomorphous alloys – Binary eutectic systems – Development of microstructure in eutectic alloys – Equilibrium diagrams having intermediate phases or components – Eutectoid and peritectic reactions – Concurrent phase transformations – ceramics and ternary phase diagrams – The Gibbs phase rule – The iron – iron carbide phase diagrams.

### Unit 2 – Nonlinear Optical Materials (11 Hrs.)

Introduction-Harmonic Generation-Second Harmonic Generation-Phase Matching-Third Harmonic Generation-Optical Mixing-Parametric Generation of Light-Self focusing of Light– nonlinear optical materials.

### Unit 3 – Ceramics (12 Hrs.)

Introduction – Glasses – Glass Ceramics – clay products – refractory's -abrasives – cements – advanced ceramics – ceramic phase diagrams – brittle fracture of ceramics – stress – strain behavior – mechanism of plastic deformation – miscellaneous mechanical consideration.

### Unit 4 – Polymers (12 Hrs.)

Introduction – Hydrocarbon molecules – polymer molecules – chemistry of polymer molecules – molecular weight – molecular shape – molecular structure – molecular configuration – thermo plastic and thermos setting polymers – copolymers – mechanical behaviour of polymers – polymer types– miscellaneous applications – advanced polymer materials – polymerization– polymer additives.

### Unit 5 - Alloys And Composites (12 Hrs.)

Introduction – Ferrous alloys: Steels – Heat treatment – Formation of pearlite – Formation of lamite – Formation of martensite – Tempering of mechanical steels. Non-ferrous alloys: copper and its alloys – aluminum and its alloys – Titanium and its alloys – Nickel and its alloys. Large particle composite – dispersion strengthened composites – influence of fiber length – influence of fiber orientation and concentration – The fiber phase – The matrix phase – polymer – matrix composites – Metal – Matrix composites – ceramic – matrix composites – carbon – carbon composites – hybrid composites – Laminar composites – sandwich panels.

### Reference Books:

1. Material Science, OP, Khanna, Dhanpat RAI, Publications, 2013.
2. Materials Science and Engineering, William D, Callister, JR, John Wiley & son, 1996.
3. Material Science and Engineering V. Raghavan, Prentice Hall
4. Balasubramanian. R., Callister's Material Science and Engineering, Wiley, India, 2010.
5. M. Arumugam - Materials Science, Anuradha Publications, Chennai
6. V. Pokropivny, R. Lohmus, I. Hussainova, A. Pokropivny, S. Vlassov. Introduction in nanomaterials and nanotechnology. University of Tartu. 2007
7. A Text book of Materials Science, P, Mani, G. Ranganath, Jayaprakash, RN, Dhanam age.

### Websites and eLearning Sources:

1. [https://onlinecourses.nptel.ac.in/noc25\\_mm19/preview](https://onlinecourses.nptel.ac.in/noc25_mm19/preview)
2. <https://youtu.be/JxQXidQmU84?si=U6qhEP2kdcVUif9b>
3. <https://youtu.be/pOEWAN00qdE?si=bnvczr5R4QzByWo5>

**COs and Bloom's Taxonomy Mapping – 26PH604**

<b>Course Outcomes</b>	<b>On successful completion of this course, students will be able to</b>	<b>BTL</b>
<b>CO1</b>	Recall and explain phase transitions, phase diagrams, Gibbs phase rule, and microstructural evolution in materials.	K1, K2
<b>CO2</b>	Apply principles of phase equilibria and alloy systems to interpret binary and ternary phase diagrams.	K3
<b>CO3</b>	Analyze nonlinear optical materials and ceramic materials including their structure, properties, and applications.	K4
<b>CO4</b>	Analyze polymer materials including molecular structure, polymerization mechanisms, and mechanical behavior.	K5
<b>CO5</b>	Evaluate and design advanced materials such as alloys and composites for engineering and technological applications.	K6

BTL (Bloom's Taxonomy Level) - K1 – Remembering, K2 – Understanding, K3- Applying, K4 – Analyse, K5- Evaluate and K6 - Create

**Relationship Matrix – 26PH604**

<b>Course Outcomes</b>	<b>Programme Outcomes (POs)</b>						<b>Programme Specific Outcomes (PSOs)</b>						<b>Mean Score of Cos</b>
	<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>	<b>PO6</b>	<b>PSO1</b>	<b>PSO2</b>	<b>PSO3</b>	<b>PSO4</b>	<b>PSO5</b>	<b>PSO6</b>	
<b>CO1</b>	3	2	1	1	1	1	3	2	1	1	1	1	1.58
<b>CO2</b>	3	3	2	2	1	1	2	3	2	2	2	1	2.00
<b>CO3</b>	3	3	3	2	1	1	2	3	3	2	2	2	2.25
<b>CO4</b>	3	3	3	2	2	1	2	3	3	2	2	2	2.33
<b>CO5</b>	3	3	3	3	2	2	3	3	3	3	2	2	2.58
<b>Total</b>													2.15

Mean Score: 3- High, 2- Medium/Moderate, 1-Low

